

EXPERIMENTAL STUDY OF HAFNIUM-SILICON PHASE DIAGRAM

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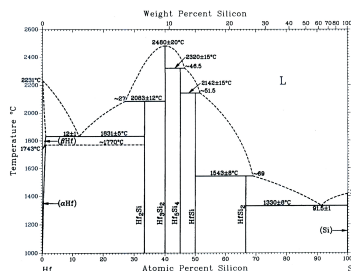
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Abstract – The currently accepted Hf–Si phase diagram indicates the stability of the phases liquid L, Hf_{SS} and Si_{SS} as well as of the silicides Hf₂Si, Hf₃Si₂, Hf₅Si₄, HfSi and HfSi₂. However, some proposals reported in the literature suggest also the stability of the phase Hf₅Si₃. The Hf–Si system have been experimentally studied in the entire composition range. Samples were prepared by arc-melting Hf (min. 99,8%) and Si (99,998%) in water-cooled copper crucible under argon. Selected samples were heat treated under argon atmosphere for 120 h at 1200°C or for 6 h at 1600°C and analysed in scanning electron microscope and X-Ray diffractometer. The results confirm the stability of the phases and the invariant transformation as proposed in the accepted phase diagram as well as the stability of the phase Hf₅Si₃.

Me-Si-B (Me – metal) alloys have been intensively studied due to their potential for the development of high temperature structural materials [1]. Considering that these materials are highly demanded in service, multicomponent based alloys seems to be the only possibility to satisfy all the requirements for structural integrity. In this sense, phase diagram information becomes extremely important. In this work, the experimental study of Hf–Si phase diagram has been carried out.

The currently accepted Hf–Si phase diagram is shown in Figure 1, based on the assessment of Gokhale e Abbaschian [2]. This diagram indicates the stability of the phases liquid L, Hf_{SS} and Si_{SS} as well as the intermediate phases Hf₂Si, Hf₃Si₂, Hf₅Si₄, HfSi and HfSi₂. However, some proposals reported in the literature suggest also the stability of the phase Hf₅Si₃ [3, 4]. Samples with compositions in all extension of the Hf–Si diagram were prepared. Mixtures of Hf (min. 99.8%) and Si (min. 99.998%) were cold pressed, arc-melted under argon atmosphere in water-cooled copper crucible with non-consumable tungsten electrode and titanium getter. The mass losses associated to the melting steps, supposing that all of them were either from Hf or Si volatilization, allowed calculate the composition adopted for each alloy, expressed by the mean value of composition interval. Selected samples were heat treated at 1200°C for 120 hours or 1600°C for 6 hours. The alloys were characterized by scanning electron microscope (SEM) in back-scattered electron mode (BSE) and X-ray diffraction (XRD) at room temperature, under Cu-K α radiation with Ni filter. For the SEM analysis, the alloys were prepared following standard metallographic procedures: hot mounting in resin; grinding in the sequence #220-#4000 SiC paper; and polishing with colloidal silica suspension (OP-S). For the XRD experiment (10° < 2 θ < 90°; 0.05° 2 θ step and 2s integration time) the samples were mechanically ground and sieved to below 80 mesh. The phases were identified based on the simulated diffractions patterns obtained from the PCW program [5] using crystallographic data compiled by Villars and Calvert [6].

The results, summarized on the Table I, Figure 2, confirm the stability of the phases and the invariant transformation proposed in the accepted phase diagram [1]. However, showed the stability of the phase Hf₅Si₃, Figure 3.



Adopted composition	Observed phases (AC)	Observed reactions (AC)	Observed phases (HT)	Originaly reactions (HT)
Hf30Si70	Hf ₅ Si ₃	L → Hf ₅ Si ₃	Hf ₅ Si ₃	L → Hf ₅ Si ₃
Hf88Si12	Hf ₅ Si ₃ , Hf ₂ Si*	L + Hf ₅ Si ₃ → Hf ₂ Si	Hf ₅ Si ₃ , Hf ₂ Si	L → Hf ₅ Si ₃ + Hf ₂ Si
Hf64Si36	Hf ₅ Si ₃ , Hf ₂ Si, Hf ₃ Si ₂ *	L + Hf ₅ Si ₃ → Hf ₃ Si ₂	Hf ₅ Si ₃ , Hf ₂ Si, Hf ₃ Si ₂	Hf ₅ Si ₃ → Hf ₂ Si + Hf ₃ Si ₂
Hf62Si38	Hf ₅ Si ₃ , Hf ₂ Si, Hf ₃ Si ₂ *	L + Hf ₅ Si ₃ → Hf ₃ Si ₂	Hf ₅ Si ₃ , Hf ₂ Si, Hf ₃ Si ₂	L + Hf ₅ Si ₃ → Hf ₂ Si + Hf ₃ Si ₂
Hf55Si45	Hf ₅ Si ₃ , Hf ₂ Si, Hf ₃ Si ₂ , HfSi	L + Hf ₅ Si ₃ → HfSi	Hf ₅ Si ₃ , Hf ₂ Si, Hf ₃ Si ₂ , HfSi	L + Hf ₅ Si ₃ → HfSi
Hf50Si50	Hf ₅ Si ₃ , HfSi, HfSi ₂	L + Hf ₅ Si ₃ → HfSi ₂	Hf ₅ Si ₃ , HfSi	L + Hf ₅ Si ₃ → HfSi ₂
Hf48Si52	HfSi ₂	L + HfSi ₂ → HfSi ₂	HfSi ₂	L + HfSi ₂ → HfSi ₂
Hf32Si68	HfSi ₂ , HfSi, Si _{SS}	L + HfSi ₂ → HfSi	HfSi, HfSi ₂ , Si _{SS} (¹)	L + HfSi ₂ → HfSi
Hf30Si70	HfSi ₂ , HfSi, Si _{SS}	L + HfSi ₂ → HfSi	HfSi, HfSi ₂ , Si _{SS} (¹)	L + HfSi ₂ → HfSi
Hf09Si91	HfSi ₂ , Si _{SS}	L → HfSi ₂ + Si _{SS}	HfSi ₂ , Si _{SS}	L → HfSi ₂ + Si _{SS}
Hf07Si93	HfSi ₂ , Si _{SS} *		HfSi ₂ , Si _{SS}	

* – primary; (1) non-equilibrium.

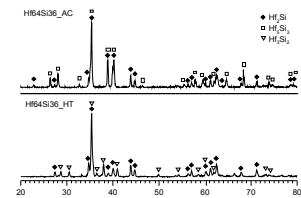


Figure 1 – The currently accepted Hf–Si phase diagram – assessment of Gokhale e Abbaschian [2].

Figure 2 – Table I – Composition, phases and reactions – observed at as cast (AC) and heat treated (HT) alloys.

Figure 3 – XRD results of the Hf64Si36 alloy as cast (AC) and heat treated (HT) at 1600°C for 6 hours.

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