## Synthesis and characterization of lithium silicates powders

Tao Tang, Zhi Zhang, De-Li Luo

China Academy of Engineering Physics, P.O.Box:919-71, Mianyang, 621900, Sichuan, P.R. China

Lithium-based ceramics, such as Li<sub>2</sub>O, LiAlO<sub>2</sub>, Li<sub>4</sub>SiO<sub>4</sub>, Li<sub>2</sub>SiO<sub>3</sub>, Li<sub>2</sub>TiO<sub>3</sub> and Li<sub>2</sub>ZrO<sub>3</sub>, have long been recognized as promising tritium breeding-materials for D-T fusion reactor blankets. Among these candidate materials, lithium orthosilicate ( $Li_4SiO_4$ ) and lithium metasilicate ( $Li_2SiO_3$ ) are recommended by many ITER research teams as the first selection for the solid tritium breeder. Li<sub>4</sub>SiO<sub>4</sub> has even been selected as the breeder material for the helium-cooled solid breeder test blanket module (HCSB TBM) in China and EU. In present study, the processes of solid-state reaction between amorphous silica and Li2CO3 powders was studied by thermogravimetry analysis-differential scanning calorimetry (TGA/DSC); the lithium silicate powders were synthesized at 700, 800 and 900 °C with Li:Si molar ratios of 0.5, 1, 2 and 4, respectively, using solid-state reaction method. The as-prepared lithium silicates were characterized by X-ray diffractometry (XRD) and scanning electron microscopy (SEM). The results show that the phase composition and morphology of the as-prepared samples change with the different synthesis conditions. At low temperature of 700 °C, all samples contain the amorphous silica, and the major crystalline phase is Li<sub>2</sub>SiO<sub>3</sub> with different microstructure for Li/Si ratio of 0.5, 1 and 2. As for Li/Si = 4, 98% purity of Li4SiO4 can be obtained at 700 °C. At high temperature of 900 °C, the significant sinterization effect will occur in all samples and Li4SiO4 will even decompose. The results also show that pure Li<sub>4</sub>SiO<sub>4</sub> can be synthesized by calcining at 800°C for 4 h, and its' solid-state reaction synthesis may be divided into two steps:

(1) 515–565 °C:  $Li_2CO_3$  +  $SiO_2 \rightarrow Li_2SiO_3$  +  $CO_2$ ;

(2) 565–754 °C:  $Li_2CO_3 + SiO_2 \rightarrow Li_2SiO_3 + CO_2$  and then  $Li_2SiO_3 + Li_2CO_3 \rightarrow Li_4SiO_4 + CO_2$ .

While Li/Si = 2, 99% purity of and pure  $Li_2SiO_3$  can be obtained at 800 and 900 °C, respectively.

**Keywords**: Test blanket module; Tritium breeder; Lithium silicates; Solid-state reaction; X-ray diffractometer; Scanning electron microscope