

Influence of the powder synthesis route in the Microstructure and Electrical Conductivity of Yttria Doped Ceria

A. S. Melo^{(1)*}, D. P. F. de Souza^(1,2)

(1) PPGCEM, Universidade Federal de São Carlos, e-mail: ariane.smelo@yahoo.com.br

(2) Depto Engenharia de Materiais, Universidade Federal de São Carlos, e-mail: dulcina@ufscar.br

* Corresponding author

Abstract –The microstructure and electrical conductivity of yttria doped ceria were analyzed in samples prepared with powder synthesized through oxide mixture and amorphous citrate. The samples prepared with powder from amorphous showed exaggerated grain growth at both sintering temperatures compared to those obtained with powder from oxide mixture. The more conductive sample was the one prepared with oxide mixture and sintered at 1600 °C.

The electrical conductivity of ceramics is strongly dependent on the purity of raw materials and the sintering conditions. Silica, a common impurity in raw materials, is usually the main limiting factor to obtain high conductivity. The microstructure and electrical conductivity of ceria based ceramics are very sensitive to the sintering temperature and the powder synthesis process. Other important factor is the appropriate choice of the dopant. Yttrium oxide has been widely used [1,2].

In this work yttria doped ceria powders were prepared by two processes: oxide mixture (PO) and amorphous citrate (PC) in order to correlate the powder characteristics with the microstructure and electrical conductivity. In the first case the mixture was prepared by vibratory mill with isopropyl alcohol for 6 hours. Subsequently the suspension was dried and the powder was deagglomerated in 80 mesh nylon sieve. The powder was calcined at 500 °C for 1 hour. After calcination, the powder was mixed with PVB in isopropilic alcohol, dried and again deagglomerated in nylon sieve 80 mesh. In the second case cerium and yttrium nitrate, both 99% purity and supplied by Aldrich, were dissolved in distilled water with citric acid and hydroxyethyl cellulose (HEC) forming a gel. The gel was dried at 300 °C generating the amorphous powder precursor that was calcined at 500 °C in oxidizing atmosphere. Pellets of the powders prepared through different routes were isostatic pressed (200 MPa) and sintered at 1500 and 1600 °C with 2 h dwell time. The sintered samples were characterized by X-ray Diffractometry (Siemens D – 5000, Cu K α 1,54Å), Scanning Electron Microscopy (SEM) (Philips XL30 TMP, Philips XL30 FEG) and Impedance Spectroscopy (IS) (HP 4192A).

The SEM analysis of samples sintered at 1500 °C, Figure 1, showed that the grain growth of sample prepared by citrate process powder (PC) was bigger than oxide mixture powder (PO). The PC samples sintered at 1500 °C showed total conductivity two times higher than the PO samples, however, become 4 times less conductive that PO sample when sintered 1600 °C. The conductivity of the grain boundary was responsible for this sharp decrease (Figure 2A and 2B).

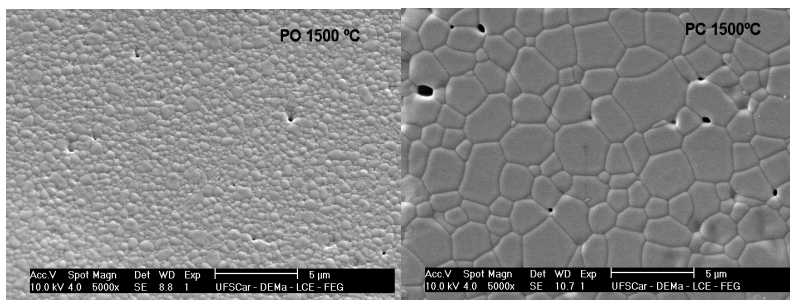
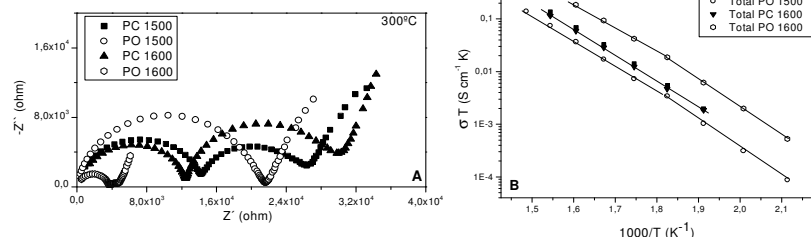


Figure 1: Microstructures of samples sintered at 1500 °C prepared with oxide mixture powder (PO) and the amorphous citrate process powder (PC).

Figure 2: Samples prepared with powders from both route and sintered at 1500 e 1600 °C: A) Electrical Impedance spectra and B) Arrhenius plot for total conductivity.



References

- [1] M. Mogensen, N. M. Sammes, G.A. Tompsett, *Solid State Ionics*, 129 (2000) 63-94
[2] V. Esposito, E. Traversa, *J. Am. Ceram. Soc.*, 91(4) (2008) 1037-1051